= DEEP SEA DRILLING PROJECT = = CLAY MINERALOGY, ADDITIONAL DATA CODED BY WHOI =

I. INTRODUCTION

A. BACKGROUND AND METHODS This data set consists of results from analyses made by a large number of scientists from various countries who applied a total of 60 different methods to clay mineral analyses.

These data were encoded at the Woods Hole Oceanographic Institution, under the direction of Dr. J. Mienert from the Initial Reports, and are intended as an extension of the x-ray mineralogy data compiled by the DSDP data management staff for Legs 1-37 of the DSDP. The data compiled at DSDP were from x-ray diffraction analyses made by the University of Califoria at Riverside X-Ray Mineralogy Laboratory, and are contained in separate data sets.

Legs 38 through 96 data were digitized by WHOI completely from published articles of DSDP volumes, ie. from a variety of different tables, bar logs, and columnar logs. The files contain both shipboard measurements and measurements made at onshore laboratories. The x-ray mineralogy data are usually presented with the relative proportions of of clay minerals (%) normalized to 100%. Purely descriptive data analyses such as a division of the relative amount of clay components into traces, rare, and common are excluded from this study because of any lack of quality control of the data. The unevenness in quality and quantity of past measurements is due to the use of different size fractions, the inadequate treatment of samples, and the inconsistency in converting XRD peak intensities to absolute or relative concentrations of clay minerals. This results in a somewhat limited treatment of deep-sea clay mineralogy. However, the data are certainly of value for studies of areas where the same method has been applied and for regional compilations of clay mineralogy data in areas of similar sediment facies.

The analysis of clay minerals is organized according to grain size, in particular in three main size fractions as follows:

Bulk	grain	sizes:	>	20	n	nic	crons	5			
Silt	grain	sizes:	<	5	,	<	10,	1-10,	2-20,	2-37	microns
Clay	grain	sizes:	<	1,	<	: 2	2 mio	crons			

The data set contains the clay minerals (%) organized in three files:

xrw_bulk.dat
xrw_silt.dat
xrw_clay.dat

The format follows the general guidelines established by the University of California at Riverside X-Ray Mineralogy Laboratory which operated from Leg 1 through Leg 37 under DSDP contract, with the following exceptions:

- 1. A section depth (top and bottom) is given for some samples.
- 2. A BSF depth is given for some samples.
- 3. Depth to top and bottom of sample interval are omitted.
- 4. The fraction field may contain micron ranges rather than the original "B", "S", or "C" for Bulk, Silt, or Clay.
- 5. Mineral mnemonic/Weight % couplets for one sample are all on one record, rather than using multiple records combined with a record position field.

II. FORMAT AND FIELD DESCRIPTIONS

A. RECORD FORMATS

The data files have a logical record length of 191 characters. The data files are sorted in the following sequence by:

Position	Format	Field Contents
=======	=====	
1-2	12	Leg ID
3-5	I3	Site
б	Al	Hole
7-9	*A3	Core
10-11	*A2	Section
12-14	**I3	Top Section Depth
15	Al	"_"
16-18	**I3	Bottom Section Depth
19-25	**F7.2	BSF Depth
26-29	A4	Fraction
30-31	12	Number of Minerals in Sample
32-191	20(A4,**F4.1)	Couplets of mineral mnemonic followed by weight percent

*For some holes, no core or section is given, only BSF depth. **These formats are >90% correct. In some cases, Top Section Depth may contain a decimal number. In some cases, weight percentages may be F4.0.

The couplets of mineral mnemonic identify the clay mineral followed by percentages. The percentages attached to the actual minerals are weight percents relative to the particular sample and fraction, but for some analyses these values have not been normalized to 100%. Despite the fact that percentages are sometimes written to 1/10 of 1 percent, the data must be viewed as semiguantative.

C. REFERENCES

For an overview of the clay mineralogy data of DSDP Volume 1-44 see G. Ross Heath, 1984. X-ray Mineralogy Studies. In G. Ross Heath (ed.), Sedimentology, Physical Properties, and Geochemistry in the Initial Reports of the Deep Sea Drilling Project Volumes 1-44: An Overview, World Data Center A for Marine Geology and Geophysics Report MGG-1, pp. 71-91.

MARC

E. X-RAY MINERAL MNEMONIC LIST:

	Four Character
Mineral	Mnemonic
======	=======
'diffuse scattering' measure	DIFF
'amorphous material' measure	AMOR
	OIIAR
MICA & MONTMORILLONITE	MICA
MONTMORILLONITE	MONT
CHLORITE	CHLO
CALCITE	CALC
KAOLINITE	KAOL.
KAODINITE K_FELDSDAP	KACH K-FF
	CUTN
	DVDT
	DADT
DALICODERITE (Atomulaito:	DALY
PALIGORSKITE (Acapuigice; Dalvoorskite & Seniolite)	PALI
AMPHIBOLE	АМРН
PHILLIPSITE	PHTL
CRISTOBALITE	CRIS
GOETHITE	GOET
MAGNETTTE	MAGN
GYPSIM	GYPS
HALTTE	HALT
ANALCITE	ANAL
HEMATITE	HEMA
ARAGONITE	ARAG
SIDERITE	SIDE
FELDSPAR (undifferentiated)	FELD
APATITE	APAT
SEPIOLITE	SEPI
RHODOCHROSITE	RHOD
MAGNESIAN CALCITE	MGCA
MIXED LAYER CLAY	MIXL
ANHYDRITE	ANHY
"variety of MONTMORILLONITE"	2-MO
ILLITE (Itmontmorillonite;	ILLI
Itmontmorillonite & Mica;	
Itmontmorillonite & Smectite)
ALKALINE FELDSPARS	AK-F
(K Feldspars; Sanidine)	
CLAY MINERALS	CMIN
ZEOLITE	ZEOL
WAIRAKITE	WAIK
SEPENTINE	SERP
PYROXENE	PYRO
PYRO-PHYLLITE	РҮРН
OPAT-CT	OPAT
NATRO-JARO	NJAR

MARCASITE

MAGHEMITE	MAGH
LEPIDOCROSITE	LEPI
HORNBLENDE	HORN

Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9 (Including the table from Heath, 1984) FRACT (MICRONS) DATA FORMAT IN IR TABLE BARLOG COLLOG BULK <2 2-20 LEG (SITES) LOCATION METH REFERENCES DEVICE SITES NOTES _____ _____ 01 1- 7 Gulf Mexico X Х M1 Rex, 1969 Brief defn of methods at Riverside lab 02 8-12 N. Atlantic X Х M1 Rex, 1970 03 13-22 S. Atlantic Х Х Х Х M1 Rex, 1970 04 Atl, Carib. Х Х Х Х М2 Rex and Murray, 1970 Riverside meth in Appendix 05 32-43 E N. Pacific X Х Х М2 Rex and Murray, 1970 06 44-60 W N. Pacific X Х Х Х М2 Rex et al., 1971 <1 Х MЗ Lisitzin, 1971 07 61- 67 W. Pacific Х Х Х Х М2 Balshaw, K.M., 1981 Х Х Μ4 Drever, J.I., 1971 Site 66 only 08 68- 75 E. Pacific Х Х Х Х Х М2 Balshaw, K.M., 1982 Semiquatitative for 2-20 microns 09 76-84 E. Pacific х х Х Х Х М2 Basov et As Leg 8 al., 1979 85-97 Gulf Mexico X 10 Х Х Х Х М2 As Leg 8; lists Beiersdorf et al., 1983

revised peak intensity vs. conc. factors Х 11 98-108 E N.Atlantic X X X Х M2 Zemmels, 1972 As Leg 8; no data site 107; no methods for Hathaway samps (bulk & <2) * Х 12 109-119 N N.Atlantic X X х х M5 Fan and no data sites Zemmels, 1972 109-110; quant. all fractions 13 120-134 Mediterranean X X Х Х Х M5 Zemmels Site 121 only** and Cook, 1973 14 135-144 N. Atlantic X X Х Х М5 Fan and Х No data Site143 Rex, 1972 new chl factor; no data 139, 142, 143.

Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9 (continued)

60 452-461 W. Pacific X M39 Desprairies, 1982 453-454 456,460-461 M37 Balshaw, Х Х 452-454 1982 61 462 W.Eq.Pacific X Х Х M32 Nagel & Х Muller, 1981 462 х х Х Х M40 Kurnosov&Shevchenko,1981 462 Х 62 463-466 N.Ce.Pacific X Х Х M40 Nagel & Schumann, 1981 463-466 Dev XRD-DRON-1. <1 <10 M41 Rateev et al., 1981 XRD-DRON-1 463-466 CuK alpha rad. @2deg/min scan speed. Verbal list relative dominance. X X,<1 X Vanek, 1981 XRD-NORELCO 463-466 ditto except M32,42 Hein & verbal list. 63 467-473 <1 х х M41 Rateev et al., 1981 XRD-DRON-1 467-473 ditto including verbal list. 64 474-481 E. Pacific X X M40 Х 477,481, Schumann&Nagel, 1982 477-479 65 482-485 E. Pacific/ ?X --- Timofeev et al., 1983 482-485 Verbal list rel. Gulf of dominance. California X X,<0.5,<1 Х M43 Rangin et al., 1983 483,485 х х Х Х Х M32 Schumann, 1983 482-485 >20 Х Х M32 Kurnosov et al., 1983 482-485 66 486-493 E. Pacific X X M32 Х 486-493 Schumann&Nagel, 1982 67 494-500 E. Pacific/ M32 Х Х Heinemann & Fuchtbauer, 494-496, Trench off 1982 499-500 Х Х Х Х M40 Kurnosov Guatemala et al., 1982 494-496, 499-500 M38 Latouche Х Х Х & Maillet, XRD-Philips 494-500 Verbal list rel.

1310 dominance. X <1 1-10 X ----Kurnosov&Shevchenko, 495,499- CuK alpha rad. 1982

Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9 (continued)

FRACT (MICRONS)DATA FORMAT IN IRLEG (SITES) LOCATIONBULK <2 2-20</td>TABLE BARLOG COLLOG METHREFERENCESDEVICESITESNOTES _____ _____

 68
 502-503
 E. Eq.
 X
 2-37

 Zimmermann, 1982
 XRD-7
 G.E.
 502
 CuK alpha rad.

 Х M44 Pacific @2 deg/min scan speed Х Х Х Х M32 502,503 CuK alpha rad. Schumann&Nagel, 1982 @1.8deg/min scan speed >20 X Х X M45 69 501, E. Eq. Х Beirsdorf&Rosch,1983 XRD-Philips 504,505 504-505 Pacific/ ADP 10 Panama Basin 70 506-510 E. Eq. Х Х Х M46 Honnorez et al.,1983 XRD-Philips 506,507 Verbal list rel. Pacific Norelco abundance CuK alpha rad. 1deg 2theta/min scan speed X,>20 X X X M47 Kurnosov et al.,1983 XRD-DRON- 506,507, CuK alpha rad. 1,5 509

71511-514S W AtlanticXXXRobert&Maillet, 1983XRD-CGR327,329,CuK alpha rad. 71 511-514 S W Atlantic Х M48 330,511, @1deg 2theta/ theta60 512,513, min scan speed 514 <1 Х M49 Varentsov et al,1983 XRD-DRON-2 511,513, CuK alpha rad. 514 @2deg/min & @ldeg/min scan speed Verbal list rel. dominance <1 <10 et al.,1983 XRD-DRON-1 511,513 CuK alpha rad. Х M49 Timofeev @2deg/min & @ldeg/min scan speed Verbal list rel. dominance 72 515-518 S W Atlantic X 2-37 M44 Zimmermann, 1983 XRD-7 G.E. 515,516, CuK alpha rad. 355,357 @2deg/min scan speed Х Х M50 XRD-RIGAKu 515,516, CuK alpha rad. Coulbourn, 1983 Miniflex 517 @2.4deg 2theta/ 2005 scan speed Verbal list rel. dominance 73 519-524 Ce.S. X X 1984 519,520, CuK alpha rad. M33 Karpoff, Atlantic 521,522, @1deg/min 523 scan speed 74 525 S E Atlantic 74525S E AtlanticXMaillet&Robert,1984XRD-CGR525,526,CuK alpha rad. Х M48

529 min scan speed

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Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9
(continued)
FRACT (MICRONS) DATA FORMAT IN IR
LEG (SITES) LOCATION BULK <2 2-20 TABLE BARLOG COLLOG METH
DEVICE SITES NOTES
_____
_____
75 530-532 S E Atlantic X
Maillot&Robert, 1984 XRD-CGR
                                                   Х
                                                        M48
                            530,532 CuK alpha rad.
theta60
                  @ldeg 2theta/
min scan speed
                           Х
                                                   X M48 Chamley
                     525,527,
et al., 1984
528,523
                                                   X M51 Stow &
                           Х
                       530
Miller, 1984
                              CuK alpha rad.
@ldeg 2theta/
min scan speed
76 533-534 Western N X
' 1023 534 Only Cretaceous
                                                   X M52 Kagami
          Atlantic/
sections
          Blake Bahama
                                            X X M48 Chamley
          Outer Ridge
                               Х
et al.,1983 XRD-Philips 534 CuK alpha rad.
1730
                  @2deg 2theata/
min scan speed
77 535-543 Gulf of
                           Х
                                             Х
                                                  Х
                                                         M48
Debrabant et al,1984 XRD-Philips 535,536, CuK alpha rad.
           Mexico
1730
          537,538, @2deg 2theta/
540 min scan speed
78 541-543 E Caribbean/
                                       Х
                                                         M53 Wright,
           XRD-Norelco --- CuK alpha rad.
1984
          Lesser
Only surface
          Antilles
sed. samples
```

intraoceanic from piston island arc cores Х X M54 Pudsey, Х Х 541,542, CuK alpha rad. 1984 543 @2deg theta/ min scan speed X X <5 Х Х M38 Latouche&Maillet, XRD-Philips 541,542, CuK alpha rad. 1984 1130 543 79 544-547 E N Atlantic/ Х Х M51 Stein&Sarnthein, XRD-Philips 544 N W African 1984 1050 Marqin Х Х Х M32 544,545, Schumann, 1984 546,547 Х х х M48 Chamley&Debrabant, XRD-Philips 544,546, 1984 1730 547 80 548-551 E N Atlantic X X Х Х M38 Chennaux et al.,1985 XRD-Philips 548,549, CuK alpha rad. 1130 550 х х Х X M55 Thiry & Pascal, 1985 548,549, 550 Х 81 552-555 E N Atlantic/ X X Х M38 Latouche & Maillet, XRD-Philips 552,553, CuK alpha rad. Rockall 1984 1310 554,555 Plateau Х Х M44 Zimmerman, 1984 XRD-7 G.E. 552,553, CuK alpha rad. 554,555 @2deg 2theata/ min scan speed

Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9 (continued) FRACT (MICRONS)DATA FORMAT IN IRLEG (SITES) LOCATIONBULK <2 2-20</td>TABLE BARLOG COLLOG METHREFERENCESDEVICESITESNOTES _____ _____ 82 556-564 Ce. N X X Х Х M38 Latouche & Maillet, XRD-Philips 558,563 CuK alpha rad. Atlantic 1985 83 504B --- Alt et E Eq. Х Х al., 1985 XRD-Philips 504 CuK alpha rad. Pacific/ Panama Basin 84 565-570 E Eq. Х Х M56 Helm, 565,567, CuK alpha rad. 1985 Pacific/off 568,569, Guatemala 570 85 571-575 Ce. Eq. NO MINERALOGY! Pacific 86 576-581 N W Pacific X X X Х Х M57 576,578, CuK alpha rad. Schoonmaker et al., 1985 581 @2deg 2theata/ min scan speed Device XRD-Philips Norelco 87 582-584 N W Pacific Х X M28,M48 Chamley 582,583, et al., 1986 584 89 585- W Eq. Pacific X Х X X M48 Chamley et al., 1986 585 90 587-594 S W Pacific <63 X M53 Gardner Х et al., 1986 591 X M48 Robert Х et al., 1986 XRD-CGR 592,593, CuK alpha rad. theta60 594 @ldeg 2theta/ min scan speed Х X M48 Stein & Robert, 1986 XRD-CGR 588,590,

theta60	591					
91				?		
92 597-602 1986	S E Pacific 59	X X 97 Verb	al list rel.		M58	Kastner,

dominance

Table 1. Summary of X-ray Mineralogical Studies Reported in DSDP Volumes 1 to 9 (continued) FRACT (MICRONS)DATA FORMAT IN IRLEG (SITES) LOCATIONBULK <2 2-20</td>TABLE BARLOG COLLOGREFERENCESDEVICESITESNOTES _____ _____ 96 614-624 Gulf of <4 X M59 Stow et al., 1986 615,620, CuK alpha rad. Mexico 621,622 @ldeg 2theta/ min scan speed х х Х M60 Thayer 618,619 Verbal list rel. et al., 1986 dominance X M52 Ishizuka Х et al.,1986 615,617, 618,619 <4 Х M59 614,619 CuK alpha rad. Pickering et al.,1986 @ldeg 2theta/ min scan speed _____ _____ A Stoffers and Muller, 1978, Zemmels et al., 1972. the Hathaway samples were analyzed at Woods Hole Oceanographic Institution. There is no information on sample preparation. Layer silicates are not differentiated in the bulk analyses. Clays (<2um) treated with ethylene glycol. Intensity vs. abundance factors not stated.

B Nesteroff (1972) refers to X-ray mineralogy done at the University of Paris, but includes no methology or data plots or listings. Composition ranges are reported in the text, but their origin (Riverside of Paris) is unclear.

C Connelly and Nalli, 1973. No methology or data listings given. Qualitative results discussed in text.

D Emelyanov et al., 1979. Carbonates, quartz, and feldspars determined for bulk sediment (using CaF2 interal standard?).

E Kossovakaya and Drits, 1978, Rateev et al., 1979, Renngarten et al., 1979, Timofeev et al., 1979, Varentsov, 1979. None of Leg 38 Russian mineralogy includes adequate methodology. All sites are assumed to have used M16 treatments.

Table 2. X-ray mineralogy: Synopsis of treatments for Legs 1-96 (refers to Table 1)

M1 - Rex, 1969. Rex's initial paper summarizes the sample treatment, intensity to concentration conversion factors, and concept of data reduction used at Riverside. The Appendix (M5) is a direct descendent of M1.

M2 - Rex reviews the data-processing scheme at Riverside. Includes filter description, intensity to concentration conversion factors, and interference corrections. The Appendix (M5) describes the Riverside procedure that eventually evolved from M2.

M3 - Lisitzin et al., 1971. Grind to <1 micron (sic), add CaF2 standard. Compare to standards for quartz and carbonate estimates. Stokes-separate < 1 micron fraction, remove carbonate with 0.1N HCl, remove oxyhydroxides with dithionite-citrate-bicarbonate (DCB), saturate with 1N MgCl2, expand with glycerine. Also use Li saturation (Greene-Kelly, 1953). Use Biscaye (1965) peak-area ratios (M11).

M4 - Drever, 1971. No information on sample treatment. M1 factors used

for abundance estimates.

M5 - Cook et al., 1974. This is the "polished" Riverside Lab methodology (see Appendix).

M6 - de Segozac, 1973. Remove carbonate by 0.1N HCl, Stokes-separate (settle and centrifuge) <2 micron, Mg-saturate. Ethylene glycol solvate? Plots of peak character and montmorillonite/illite and chlorite/illite (4.7/5A) peak ratios.

M7 - von Rad and Rosch, 1972. Bulk samples, formic-acid-treated fractions run if carbonate rich. Abundance estimates as "main", "abundant", "common", and "tract": (>40, 20-50, 10-20, 3-10%). Columnar logs with abundance symbols.

M8 - Roberson, 1973. Wash out salt, disaggregate ultrasonically, Stokes-separate <2 micron by centrifuge, pipette on glass. Use "similar to" Johns, et al. (1954) abundance estimates.

M9 - Hayes, 1973. Stokes-separate <2 microns (centrifuge), Mg-saturate, ethylene glycol solvate. Intensity to conversion ratios - mica/ montmorillonite:illite:chlorite = 1:3:3. Separate chlorite from vermiculite by 7A/14A ratio (2 to 0.2, respectively).

M10- Okada and Tomita, 1973. Air-dry aggregates on glass, expand with ethylene glycol. Intensity to conversion factors for montmorillonite:-illite:kaolinite = 1/2.6:1:1/1.2 (based on matching standards to unknowns). Size fraction not stated, probably bulk.

M11- Venkatarathnam, 1974. Biscaye (1965) methodology and intensity to abundance factors. Sodium acetate (pH=5) carbonate removal, DCB oxyhydroxide removal, sodium carbonate disaggregation, then size separate. Paste smear <2 micron fraction on glass, ethylene glycol solvate. Conversion factors - montmorillonite:illite:kaolinite: chlorite = 1:4:2:2. Gerbuneva, 1976. Uses < 1 micron, no DCB. Zimmerman, 1977. Uses <4 micron.</pre>

M12- Gostin and Moriarity, 1975. Carbonate removal by glacial acetic acid, Stokes-separate < 2 micron fraction, NaOH remove opal, DCB remove oxyhydroxides, Mg saturate, glycerine solvate. Conversion factors - montmorillonite:illite:kaolinite:chlorite = 1:4:2:4. Also uses CEC (Ba) to cross check montmorillonite.

M13- Perry et al., 1976. Stokes-separate < 1 micron, < 1 micron, pipette on glass. Ethylene glycol solvate? Use relative peak intensities (no abundance conversion).

M14- Eslinger and Savin, 1976. Stokes-separate <0.3, 0.3-0.7, and >0.7 microns. Ethylene glycol solvate. Illite/smectite:illite:kaolinite:- chlorite abundances given by I/S 003/005-1/2 I 001:I 001:K 002:C 004.

M15- Kastner, 1976. Stokes-separate <2 micron, >2 micron. Use Drever's (1973) technique. Data shown as relative peak intensities (illite 10A:chlorite 7A:illite/smectite 17A).

M16- Kessovskaya and Drits, 1978. Stokes-separate <1 and 1-10 micron. Other methodology not stated. Mentions glycerine saturation. Description and rough estimates of clay mineralogy. (Method is also described in Rateev et al., 1979, McCoy et al., 1977, Timofeev et al., 1979, and Varentsov, 1979).

M17- Perry et al., 1979. Stokes-separate <1 micron, pipette on glass,

ethylene glycol solvate. Use Biscaye (1965) intensity to concentration factors (M11).

M18- White, 1976. Grind dry, then wet. Stokes-separate <5 micron. Ethylene glycol solvate. Use Biscaye (1965) factors (M11). Sites 338, 340, 341, 343, 344, 345.

M19- White, 1976. Bulk: wash, dry crush, report as "present", "abundant", "major". Clays: Stokes-separate <2 micron, vacuum mount on tile, ethylene glycol solvate. Use Biscaye (1965) factors (M11). Sites 346 to 349.

M20- Matsumoto, 1978. Bulk: dry grind. No intensity to concentration conversion factors. Clays: <2 micron, ethylene glycol solvate, hydrazine hydrate expand kaolinite. Intensity to abundance conversion factors - smectite:mica:kaolinite:chlorite + 1:1.7:1:1.5.

M21- Siesser and Bremner, 1978. Dialyse 24 hr. Bulk: dry, crush, press, tabulate peak intensities above background. Clays: remove carbonate with 25% acetic acid, Mg-saturate, pipette on glass (warm to dry), ethylene glycol solvate. Use Johns, et al. (1954) factors (kaolinite and chlorite undifferentiated).

M22- Couture, 1978. Glass mount. Remove carbonate by pH=5 acetate buffer, ethylene glycol solvate. Data listed as "present", "abundant", "dominant".

M23- Melieres, 1978. Bulk samples. Use NaF internal standard, ethylene glycol solvate. Abundances estimated by comparison with standards. For clays, remove carbonte with 10% HCl, pipette on glass, ethylene glycol solvate. Montmorillonite:illite:kaolinite:chlorite = 1 (14A):1(10A):1:1 (7A divided by 003:006 ratio).

M24- Trimonis et al., 1978. For clays, Stokes-separate <2 micron, remove carbonate with 1N HCl, Mg-saturate, glycerine solvate. Abundance from Biscaye (1965) factors (M11).

M25- Koch and Rothe, 1979. Bulk plus <2 micron. Carbonate removed from clays with 10% HCl, pipette on glass (checked against smear mount). Clay ratios from Biscaye (1965) factors (M11). Bulk results comparable to Hathaway's (see footnote a, Table 1).

M26- Pastouret et al., 1978. No methodology or accessible reference. Qualitative data (abundant, common, present?) for 20-63 microns. Percentages (conversion factors not stated) for clay minerals.

M27- Timofeev et al., 1979. Carbonate removal by acetic acid. Clay minerals were studied in the fraction <10 microns. Treatment with glycerine solvate. Verbal listing of clay minerals.

M28- Chamley and d'Argoud, 1979. Carbonate removal by 5N hydrochloric acid. Decantation of <2 micron fraction by using Stokes law. Semiquantitative evaluations are based on the peak heights and areas (after Chamley, 1971). Relative error +/- 5%. Heights of 001 illite and chlorite peaks are taken as references.

M29- Latouche, G. and Maillet, 1979. Carbonate removal by N/10 hydrochloric acid. Ethylene glycol treatment before analysis. Semiquantitative estimates from the height (? and areas) in the diagrams of glycolated slides. The height of 001 peaks was used to determine the percentage of smectite (17A), illite (10A), and kaolinite + chlorite (7.1A).

M30- Timofeev et al., 1979. Carbonate removal by N/10 hydrochloric acid. Samples were measured natural, saturated with glycerine, and heated at 550 degrees C for identifying mixed-layer minerals of the Ch-M type.

M31- CEPM Laboratory, 1980. Carbonate-free fraction of < (?) 5mm. The 1 day mineral content were estimated in relative percentages from the heights and the areas of the x-ray diffraction peaks. No further details.

M32- Mann and Muller, 1980. Clay minerals in the <2 micron fraction were x-rayed in an untreated state, glycolated state, and when necessary, after heating. Biscaye's method (1965) was used to determine the clay minerals by multiplying the peak (peak 2 ?? (in degrees):smectite + 5.2 (17A), chlorite = 12.3 (7.2A), illite = 8.8 (10A) kaolinite + 12.3 (7.2A), polygorskite = 8.4 (10.5A), talc = 3.4 (9.3A), sepiolite = 7.4 (12.0A) area by factors (smectite = 1, chlorite = 2, illite = 4, kaolinite = 2, polygorskite = 1, talc = 1, sepiolite = 1). The sum of the clay minerals was 100 percent. The kaolinite/chlorite peak at 2 ?? = 12.3 degrees was divided by peak splitting.

For the bulk mineralogy the peak heights are multiplied by factors (REF.101). When multiplied they add up to 100%. However, if the amount of amorphous material, i.e., volcanic glass, opal, and amorphous clay minerals, vary, the factors have to be redetermined. These factors apply only for minerals which are very similar to those from Site 502 and 503.

M33- Karpoff et al., 1980. Identification of clay minerals was made on three types of oriented aggregates: untreated, ethylene-glycol-treated, and heated (Mise au point collective, 1975).

M34- Mann and Muller, 1980c. Method of M11 slightly modified because of the high amount of amorphous constituents in the sediments. The amorphous constituents influence the (020) peak in such a way that its base line cannot be exactly traced. Thus, it is impossible to quatify the percentage of clay minerals by direct peak measurements. The method applied is to subtract all other consituents that can be determined by x-ray methods. The margin of the error of this method is all the larger the greater the amount of amorphous constituents.

M35- Kurnosov et al., 1980. Carbonate removal by N/10 hydrochloric acid. Decantation of <1 micron and 1-10 micron fraction. Samples were dried, treated with ethylene glycol, and heated at 500 to 550 degrees C. Some samples were K-saturated to test the nature of expanding minerals (Weaver, 1968). Relative proportions of clay minerals were analyzed by using Biscaye's method (1965). Ref. 125, 126, 127, 129.

M36- Aoyagi and Kazama, 1980. Method of analysis is explained in detail by Oinuma and Kobayashi (1966) and Aoyagi (1967). X-ray diffraction peaks of each mineral of powdered sample measured are 001 for montmorillonite, illite, and kaolinite, 002 for chlorite and plagioclase, 020 for clinoptilolite and gypsum, 101 for quartz, 104 for calcite, dolomite, and siderite, 110 for hornblende, and 200 for pyrite and halite. Existence of illite-montmorillonite mixed-layer minerals is inferred from the width of the 17A peak of glycolated samples.

M37- Balshaw, 1981. Carbonate removal by 0.1N (pH 5) sodium acetate and acetic acid. 0.2M ammonium oxalate and 0.2M oxalic acid were used to dissolve amorphous Fe-hydroxides and oxyhydroxides. The <2 micron fraction was separated by settling using Stoke's law. Deionized water and Calgon were used to minimize flocculation of the clay particles.

The resulting clay was treated with a 1.0M solution of Mg Cl to

saturate the clay exchanges with Mg ions and finally washed. The slides were x-rayed after air drying and again after treatment with ethylene glycol. The glycolation caused a shift of the smectite peak from 12-14.0A to 18.0A. The illite peak at 10.0A, and the chlorite and kaolinite peaks which are coincident at 7.0A are not affected by glycolation. The relative percentages of the clay minerals were calculated by using techniques outlined by Biscaye (1965).

M38- Latouche et al., 1982. Pulverized sediments of bulk samples were analyzed according to the powder diffractogram method. Semiquantitative analysis of the content of quartz, calcite, and feldspars is based on their diffraction peak height which is compared with peaks of mixed synthetic reference samples. Carbonate removal of the clay fraction (<2 micron) by N/10 HCl. The <2 micron fractions were separated by gravity settling. The slides were x-rayed after air drying and then after heating at 550 degrees C for one hour. The second was scanned after treatment with ethylene glycol. Identification of clay minerals after Brown (1961) and Thores (1975). Ref. 167 slightly changed method by: (A) using 10N HCl. The clay fraction abundance, evaluated on the basis of peak heights, has an estimated experimental error of +/- 10%.

M39- Desprairies, 1982. Clay minerals in the carbonate free <2 micron fraction were determined by various treatments (heating, glycol, hydrazine). The relative abundances of the principal clay minerals were estimated by the percentage of the basal peak intensities rather than by the percentage of the basal peak areas.

M40- Kurnosov and Shevchenko, 1981. Identification of clay minerals in the wire fraction <2 micron and 2-20 micron using the method of Biscaye (1964) (see also M11). Conversion factors montmorillonite:illite: kaolinite:chlorite = 1:4:2:2.

M41- Rateev et al., 1981. Samples (? carbonate free) were prepared for the <1 micron and partly for the <10 micron fraction in three states: air dried, ethylene-glycol-treated, and heated at 550 degrees C. Kaolinite is identified by standard peaks at 7.15 and 3.57A, which are disappearing after heating at 550 degrees C and preserved after treatment with 10% HCl. Montmorillonite has peaks from 14.1 to 14.7A in an air dried state which expand to 17.9A after saturation with glycerine. Montmorillonite-illite mineral mixed layer clay is characterized by an asymmetrical peak at 14.7A in an air dried state, and at 18.8 to 19.6A after saturation with glycerine.

M42- Hein and Vanek, 1981. Carbonate and organic matter removal by sodium acetate and acetic acid. The <2 micron fraction was separated by centrifugation. The fraction was x-rayed after Mg saturation and glycolation. Peak areas were used to calculate relative amounts of clay minerals (after Biscaye, 1965).

M43- Rangin et al., 1983. The <2 micron, <0.5 micron, and <0.1 micron fractions were separated by centrifugation. The ratio of the areas under the main phyllosilicate peaks indicate the clay mineralogy (? no abundance conversion).

M44- Zimmerman, 1982. Carbonate removal by 0.6N acetic acid. Separation of the 2-37 micron and <2 micron by calculated settling times. Slides were x-rayed before and after glycol solvation. Chlorite and kaolinite abundance is indicated by the area under the 7.1A peak. The separation of these two minerals is based on the slow-scan method of Biscaye (1964). Illite has a well defined peak at 10A that was unaffected by ethylene glycol solvation. A broad peak at 17A after solvation is assigned to smectite (montmorillonite). The conversion factors of Biscaye (1965) (see M11) are used: montmorillonite:illite:kaolinite: chlorite = 1:4:2:2.

M45- Beiersdorf and Rosch, 1983. Clay minerals of the non-CaCO3 components in the grain size fraction <2 microns, 2 to 20 microns, and >20 microns were determined qualitatively from glycolated powder samples. The concentration was estimated from peak height measurements which were compared with those of pure standards.

The concentration of all other components - especially feldspar and smectite, together with the amorphous scatter - were empirically estimated and balanced to 100%.

M46- Honnorez et al., 1983. Samples of air dried, ethylene-glycol treated, and heated smear slides of the <2 micron and bulk sediment fraction were x-rayed. Some samples were analyzed by subdividing the <2 micron fraction in arbitrary fractions between 2 and 0.4 microns. The XRD measurements did not indicate any difference and their clay mineralogies were similar to that of the bulk sample.

M47- Kurnosov et al., 1983. Samples of air dried, ethylene-glycol treated, and heated (500 to 550 degrees C for 1 hour) smear slides of the <2 micron, 2-20 micron, and bulk sediment fraction were x-rayed.

M48- Robert and Maillot, 1983. Carbonate removal by 0.2N hydrochloric acid in the <63 micron sediment fraction. The <2 micron fraction was separated by settling using Stoke's law. Oriented aggregates were made on glass slides. The untreated sample, the glycolated sample, and the sample heated for 2 hours at 490 degrees C were x-rayed. Semiquantitative evaluations were based on the peak heights and areas (after Chamley, 1971). The height of the 001 illite peak of the glycolated sample was taken as a reference. Compared to this value, smectite, attapulgite, and irregular mixed-layer clays were corrected by multiplying their peak height by a factor of 1.5 to 2.5. Well-crystallized kaolinite was corrected using a factor of 0.5. The relative error is +/-5%. (comparable to M28).

M49- Varentsov et al., 1983. The fraction < ? 1 micron had been air dried, treated with glycerine, and heated at 550 degrees C. The interpretation of these x-rayed samples are based on comparison of the experimental data with corresponding models of Drits and Sakharov (1976).

M50- Coulbourn, 1983. Bulk samples have been x-rayed. Diffractograms of standards were compared with diffractograms of core samples. Samples were not treated with ethylene-glycol and run a second time to distinguish chlorite from montmorillonite. The quantification of diffractograms attempted is an approximate version of a method of mutual standards (Rex, 1970).

M51- Stow and Miller, 1984. The <2 micron fraction was separated by ultrasonic treatment and settling and x-rayed for untreated and glycolated samples. Semiquantitative analysis of the clay mineral diffractograms are based on the peak areas of glycolated samples (after Biscaye, 1965). The weighting factors are: 2 (kaolinite-chlorite), 4 (illite), 1 (smectite), and 1 (mixed layers). REF. 168: Carbonate free <2 micron fraction was analyzed (Lenge, 1982).

M52- Kagami et al., 1983. The <2 micron fraction was collected by sedimentation (no further details) and x-rayed. Quantitative and qualitative estimations are based on the method by Sudo et al. (1961) and Oinuma (1968).

M53- Chamley et al., 1983. Preparation and analysis of the samples after the method by Hein et al. (1976). Clay percentages were determined on the diffractogram of the glycolated samples by the peak area method and weighted by the conversion factors after Biscaye (1965).

M54- Pudsey, 1984. Bulk samples and the <2 micron sediment fraction have been x-rayed. The <2 micron fraction was settled in sodium hexametaphosphate suspension to prevent flocculation of sediments. Each sample of the <2 micron fraction was x-rayed after (1) saturation with ethylene glycol and (2) after heating to 500 degrees C. Peak heights and areas are measured in diffractograms and are used to obtain a semiquantitative estimate of the mineralogy. The estimation of the mineralogy in percent is based on the measured peak heights and areas times a combination of weighting factors given by Rex and Murray (1970), Biscaye (1965), and Mann and Muller (1980).

M55- Thiry and Pascal, 1985. Bulk samples were treated with 1/10N HCl. The <2 micron fraction was separated by settling, dried, saturated with ethylene glycol and hydrazine hydrate, and later heated at 490 degrees C. Clay minerals and mixed layers (%) were determined by their peak areas in the diffractogram of the samples treated with ethylene glycol.

M56- Helin, 1985. No carbonate removal. The fraction <63 microns was washed several times using 0.01N NH40H. The fractions <2 microns, 2 to 6.3 microns, 6.3 to 20 microns, and 20 to 63 microns were collected by Atterberg separation after Stoke's law (Muller, 1964). The <2 micron fraction of the air dried samples, the heated samples at 500 degrees C, and the glycolated samples were analyzed. The clay mineral distribution in percent (<2 micron) was determined by using Biscaye's method (1965). The percentages of nonexpandable layers in smectites were calculated by using Reynolds and Hower (1970) and Brindley and Brown's (1980) methods. The partial destruction of the 7-A peak after heating was interpreted as indicating the presence of chlorite. The crystallinity of smectites was investigated in untreated and glycolated samples by measuring the peak height/width at half height ratio.

M57- Schoonmaker et al., 1985. Organic matter was removed by treatement of bulk samples with 5% sodium hypochlorite buffered to pH 9.5 with HCl (after Anderson, 1963). The <2 micron fraction was then separated by centrifugation. Samples were dried at room temperature, treated with ethylene glycol, and an additional aliquot of the <2 micron fraction was treated with Na citrate-Na dithionite to remove amorphous Fe-oxyhdroxides (after Mehra and Jackson, 1960). Clay-mineral abundances were calculated from peak areas by using the "triangle method" (Mann and Fischer, 1982). The percentages of the clay minerals were determined by using weighting factors (Biscaye, 1965), Mann and Muller (1980) for the peak areas. In contrast, the abundance of mixed layers is determined by using peak heights instead of areas (after Hoffman, 1976).

M58- Kastner, 1986. The <2 micron fraction of the bulk sample and that of the >2 micron fraction of the insoluble residue were determined by the Stoke's law settling method. XRD analyses were used to establish a crystallinity index for goethite.

M59- Stow, 1986. The <4 micron fraction was separated by settling. Peak heights were measured for all minerals and normalized to 100% without any application of weighting factors.

M60- Thayer et al., 1986. The bulk sample and the <2 micron fraction were x-rayed. XRD measurements for bulk sediments utilized the technique of Roberts (1982), while the <2 micron fraction was analyzed in a commercial laboratory (without further details).

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This data set was received from the Woods Hole Oceanographic Institution combined with a copy of the x-ray mineralogy data set compiled by the DSDP for Legs 1-37. The two sets of data contained too many differences to be left as a combined set, and so were split into the original DSDP files and into the new WHOI files.

Some differences included:

1) Original DSDP data contain relative terms rather than percentages in some cases (trace, abundant, etc.), 2) The fields giving depth to sample, section, or BSF did not overlap between the two data sets, 3) The DSDP data were generated under fairly uniform conditions, all by the University of California at Riverside X-Ray Mineralogy Laboratory, and retain more of their uniform value when separable from the additional analyses (Note that some data prior to Leg 38 were encoded by WHOI).

Some modifications were made to the WHOI data by NGDC: 1) Leg ID was shortened from 3 to 2 digits to provide uniformity with all of the other DSDP sediment and hardrock data sets. 2) Duplicate 4-digit site and hole fields were reduced to a single 3-digit site and 1-digit hole (blank, A, B, etc.) to provide uniformity with all of the other DSDP sediment and hardrock data sets. 3) An alpha 7-digit Section Depth was justified into 2 separate depth fields separated by a "-" to allow numeric searching on the field. 4) When "CC" was found in the Section Depth field sporadically, it was moved to the section field to be consistent with labelling in other parts of the WHOI data set and with other DSDP data sets. 5) Hole assignments conflicting with the DSDP Coredepths file were checked against the Initial Reports and changed where neccessary. We advise caution, however, in assuming that all holes cited are now correct, because only those conditions (i.e. out of range core # for a hole) which caused a mis-match with the Coredepths file were corrected. Other errors which did not cause a mis-match may still exist. These data have not undergone the thorough quality control that data sets compiled by the DSDP data group have been through.